

26 September - 2 October 2010, Carqueiranne, France

Incommensurately modulated crystals

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Incommensurately modulated crystals and applications

Examples of modulated structures

- $Na₂CO₃$
- A member of the scheelite family, $KSm(M_0O_4)_2$
- Extension of the structure type concept exemplified with the scheelite structures
	- Applications: structure property relations
- The complex structure of metallic elements
	- Barium, Rubidium
	- Gallium II (2.8 GPa)

Acta Cryst. (1964). 17, 614

An anomaly in the crystal structure of Na₂CO₃. By ELLY BROUNS and J. W. VISSER, Technische Physische Dienst $T.N.O.-T.H.,$ and $P.M.$ DE WOLFF, Technische Hogeschool, Delft, The Netherlands

> In the course of routine investigations we tried to index the powder pattern of anhydrous sodium carbonate. This proved to be uncommonly difficult. A clue to the solution was eventually obtained through application of a computer method (de Wolff 1963). This led to a C centered monoclinic unit cell (Table 1) which, however, did not explain all powder lines. A considerable number (roughly 20% of the total) of lines, mostly of weak intensity, withstood every attempt either to incorporate them on the basis of a multiple cell, or to eliminate them by different ways of chemical preparation.

Table 1. Crystallographic data of Na₂CO₂ at various temperatures

The phases of Na_2CO_3

(Dušek et al., Acta Cryst. 2003 Arakcheeva et al. Acta Cryst. 2005)

γ -Na₂CO₃. Projection of the reciprocal space along c^*

As is often the case in diffraction, the examination of the reciprocal space reveal the presence of twins or domains.

In the present case, we observe the presence of three domains.

 γ -Na₂CO₃. The importance of ADP modulations

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 F_o for atom 01 ΔF map without ΔF map with ADP ADP modulation modulation for for atom O1 atom O1 $x_3 = 0.285$, $x_1 = 0.102$ $x3=0.285,x1=0.102$ X_4 X_4 **CO** 0.5 0.5 0.5 0.0 0.0 0.0 0.40 $\overrightarrow{x_2}$ 0.15 $0.40 \frac{1}{12}$ 0.15 0.15 $0.40 \overline{(b)}$ (a) (c)

 $\overline{x_2}$

Structure description of $Na₂CO₃$ (low temperature phase)

Na-O distance Histogram

Na₂CO₃. The role of the m_M and m_v planes in the incommensurate γ phase.

$Na₂CO₃$. What can you find in an incommensurate structure ?

Analogy of the modulation wave and the temperature dependence in the ${110}_{hex}$ family of planes.

Solid lines: interatomic contact distances < 3.1 Å

$Na₂CO₃$. The phase transition mechanisms

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$Na₂CO₃$. The phase transition mechanisms

Evolution of the Na environment in the vicinity of the C atom.

The coordination number (CN) of the C atom is limited to the C—Na distances $<$ 3.1 Å

International School on Aperiodic Crystals 27/09/10 What did we learn from the incommensurate nature of structure ?

- The structure of $Na₂CO₃$ would be better represented by the chemical formula ${\sf Na}_{\rm 4/3}$ Na $_{\rm 2/3}$ CO $_{\rm 3}$ indicating the two different natures of the Na atoms
- The series of phase transitions can be much better understood in terms of next of next-next nearest neighbours
- Incommensurate structures contain structural details which can be found in other temperature modifications.
- The incommensurate character of the structure is the direct consequence of subtle chemical interactions which can be exploited in simulations of modelling.

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The incommensurately modulated structure of a scheelite: KSm(MoO₄)₂ (Arakcheeva et al. 2008)

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Electron diffraction pattern of $KSm(Moo₄)₂$

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Structure refinement of the modulated structure of $KSm(MoO₄)₂$ by synchrotron powder diffraction

Occupation modulation functions and corresponding displacive modulations of Sm and K

The modulation functions can be selected from a pool of basic functions (harmonics, crenel, saw tooth and others. They can also be combined.

Here, the occupation function of Sm and K results from a combination of crenel and harmonics.

Occupation modulation functions and corresponding displacive modulations of Sm and K

Here, the occupation function of Sm and K are modelled with harmonics. Occupation and ΔF sections shows some overlap.

Criteria for the selection of the best model

The Crenel and the harmonic models do not perfectly fit. There are still some residual peaks on the lower part of the powder diffraction diagrams.

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Criteria for the selection of the best model

The vicinity of the $A = (K_{1/2}Sm_{1/2})$ position in the structural model with randomly distributed K and Sm atoms.

The sections represent electron density (top) and residual electron density (bottom) maps.

The model must obviously be improved to decrease the residual electron density.

Side question: What can we deduce from the shape of the three sections ?

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And the final results…

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 $Sm[MoO₄]$ $K[MOO₄]$

Additional periodicity given by the vector // q with magnitude 1/q

Comparison of the HREM image and corresponding Fourier transform with the X-ray model

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Extension of superspace for the description of families of compounds

The model of superspace gives also a very efficient method to describe families of compounds with common building blocks (also called modular structures). We use the property of varying the modulation vector and the variable t.

A single $(3+1)$ D model *e.g.* can thus generate a multiplicity of structures, some commensurate and some incommensurate. We shall exploit this property with examples from the family of sheelite structures.

The scheelite family $\left(A',\,A''\right)_{\mathsf{n}\text{-}\mathfrak{\delta} A} \!\!\left[(X\,{}',X\,{}'')\mathsf{O}_4\right]_{\mathsf{n}\text{-}\mathfrak{\delta} X}$ (Arakcheeva et al. 2008)

The occupation functions of various Scheelite

A single superspace model describes the family. Each member is characterised by the modulation vector, occupation and modulation functions of the independent atoms.

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Choice of q-vectors for the 3D members

 $RbBi[MoO₄]$ ₂ K₂Th[MoO₄]₃ Bi₃ [FeO₄][MoO₄]₂ Na₄Zr[MoO₄]₄

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The $(A',A'')[(X',X'')]$ $(3+1)D$ Scheelite family

One size fits all!

Illustration of the concept with a few examples.

Superspace group: $12/b(\alpha\beta 0)00$

Basic structure: **Scheelite**

Variables: $q = \alpha a^* + \beta b^*$ \mathcal{R}_{I} occupation functions of A and X

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$K₂Th[MoO4]₃$

Projection on the ab-plane

Occupation function $q = 0a^* + \frac{2}{3}b^*$

$Bi₂[MoO₄]₃$

 $q = \frac{2}{3}a^* + \frac{1}{3}b^*$

Occupation function

$\textsf{Na}_4\textsf{Zr}[\Box\textsf{MoO}_4]_4$

The q-vector defines the wave of composition

36

 l_2

h

C

 \bigcap

 \circ

 \bigcirc

 \bullet C

q

 \bigcap \bullet

 \bigcirc \circ \bullet

 \circ \bigcirc \bigcap

 \bigcirc

k

The q-vector defines the wave of composition

Application of the superspace concept: structure property relations (luminescence)

(A. Arakcheeva, private communication)

Most of the Scheelite structures listed in the PDF (ICDD) database are average structures. A large number of them are incommensurately modulated.

The presence of satellites reflections are barely recognisable on conventional powder diagrams. They are evident from synchrotron data.

The resolution of their structure in superspace shed some new light on their structure property relation.

Some more with structure property relations (luminescence)

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Some more with structure property relations (luminescence)

Superspace and structure-property relations

- The superspace concept is particularly efficient in order to reveal fine details in complex structures
- Neglecting to account for "small" peaks in diffraction patterns or by considering them as "impurities" is tantamount to average the structure and consequently disregard all fine details of the structure.
- The relation between structure and properties has often been overseen due to the fact that the superspace character of the structures was not recognised.

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The structures of metals under high pressure

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Often exhibit incommensuration by forming structures of the hostguest type. In particular self-hosting structures in the case of elements.

Ba^{IV} exists in two variants, tetragonal (top) and monoclinic (bottom)

Ba-Ba distances in Ba^{IV}

Many more examples of incommensurate structures of metallic elements exists

 α -U < 37 K ? P2/m11($\frac{1}{2}$ β γ) ? Marmeggi et al. 1982 van Smaalen et al. 1987 $q=(\frac{1}{2}, .176, .182)$ Main modulation along a

Structure modulations in metals under high pressure

PRL 97, 115501 (2006)

PHYSICAL REVIEW LETTERS

week ending **15 SEPTEMBER 200**

Apparently Complex High-Pressure Phase of Gallium as a Simple Modulated Structure

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²Institute of Geosciences, Crystallography, University of Kiel, 24098 Kiel, Germany ³Department of Physics, University of Picardie, 80000 Amiens, France (Received 4 July 2006; published 13 September 2006)

The phase of gallium GaII, with symmetry $C222₁$ and 104 atoms per unit cell, has been recently reported as an example of *structural complexity* under high pressure. It is shown here that this phase is a simple modulated distortion of an average structure of Fddd symmetry with all atoms structurally equivalent. The modulation can be described with only 4 parameters and satisfies symmetry properties described by a centrosymmetric superspace group. The structural distortion is dominated by a frozen transversal mode associated with a single irreducible representation of Fddd, with a wave vector on the line Q , at an edge of the Brillouin zone. The average structure can be related with an hep configuration through simple sliding of hcp layers, reminiscent of the hcp-bcc Bürgers mechanism.

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Gallium II (Degtyareva et al. 2004)

TABLE I. Refined atomic coordinates of Ga-II ($oC104$) at $2.8(1)$ GPa and RT in space group $C222₁$. The refined lattice parameters are $a = 5.976(1)$ Å, $b = 8.576(1)$ Å, $c =$ $35.758(3)$ Å. The number of nearest-neighbor (nn) distances is given in ranges (i) $2.65-2.7$ Å, (ii) $2.7-3.1$ Å, and (iii) $3.1-$ 3.3 Å. These ranges are selected for comparison with Cs-III and Rb-III [7,8], and the upper bound of 3.3 Å is set at the first minimum in the distribution of contact distances.

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Gallium II (Perez-Mato et al. (2006)

Analysis of the "complex" structure

- four hexagonal layers normal to b.
- Two are related by the C centring.
- The two independent layers (a) and (b) looks very similar.

Construction of an ideal structure

• Smaller cell $c_{\text{aver}}=c/13$ (c).

Could this structure be interpreted as a modulated structure?

Gallium II (Perez-Mato et al. (2006)

Atomic displacement in relative units

- along (a) , (b) and (c)
- grey and black points are related to the two independent layers
- fit the same type of modulation
- $u_x(x_4) = A\cos 2\pi x_4 + B\cos 6\pi x_4$
- $u_v(x_4) = Asin2\pi x_4$
- $u_z(x_4) = Asin4\pi x_4$
- $q = n/13$ c^* _{aver}, best choice with $n=9$

The structure is better described as a modulated structure in superspace group *Fddd*(00γ)0s0 with <u>4</u> independent parameters (instead of 38 in the original paper.

Questions

